

I. Product and Company Identification

1.1 Identification of the substance or preparation:

OCS SODIUM CHLORITE 25%

1.2 Company Identification: Odour Control Systems Limited

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Hawarden Industrial Park

Manor Lane, Hawarden, Deeside. CH5 3PP.

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II. Composition/Information on Ingredients

2.1 Chemical Composition: Preparation of proprietary oxidising ingredients including Aqueous

solution of stabilised buffered chlorine dioxide.

2.2 Risk Phrases: R8 Contact with combustible materials may cause fire.

R22 Harmful if swallowed.

R32 Contact with acids liberates toxic gas.

2.3 Classification/Symbol: O Oxidising

XN Harmful

III. Hazards Identification

3.1 Adverse Health Effects: Irritating to eyes.

Harmful if swallowed.

3.2 Environmental Effects: Toxic to aquatic organisms.

3.3 Physical & Chemical In contact with acids or when heated or under sunlight, may develop very

toxic gas (chlorine dioxide) which may cause risk of explosion. Contact with combustible materials (grease, fats, wood, cellulose,

paper, etc) may cause fire.

IV. First Aid Measures

4.1 Eye Contact: Immediately flush eyes with water for at least 15 minutes.

Seek medical advice if symptoms persist

4.2 Skin contact: Remove contaminated clothing and wash affected areas with soap & water.

Soak contaminated clothing with water. Seek medical attention if irritation persists.

4.3 Inhalation: Remove to fresh air. Seek medical attention. (If the patient is affected by

chlorine dioxide, this is very toxic and requires urgent medical attention).

4.4 Ingestion: Rinse mouth with water and give water or milk to drink.

Seek urgent medical attention.

4.5 Immediate Medical Advice Treat symptoms. Risk of pulmonary oedema.

V. Fire Fighting Measures

5.1 Special Fire/Explosion May decompose in fire producing toxic chlorine compounds.

Hazard: Strong oxidising agent - will assist combustion.

5.2 Products of Combustion: Not combustible. Toxic chlorine compounds maybe released in fire.

5.3 Fire Fighting Procedures Wear S.C.B.A. for chlorine/chlorine dioxide.

Extinguishing Media: Keep containers cool with water spray.

Avoid dispersion in the water courses.

Do not use carbon dioxide or organic material.

In case of spreading, avoid drying by dilution with plenty of water.

VI. Accidental Release Measures

6.1 Personal Protection: Wear goggles, plastic gloves (not rubber) and boots.

Eyewash facilities should be available.

6.2 Environmental Precautions Do not allow to dry. If possible, drench with water.

Contain with inert material.

Pump into a suitable container or otherwise absorb with non combustible materials. If substance has entered a water course or sewer advise

relevant local authorities.

6.3 Methods of Cleaning Up: Flush away any residues with excess water.

Spillages greater than 25 litres should be decontaminated with sodium sulphite solution with a contact time of at least 10 minutes before flushing.

VII. Handling and Storage

7.1 Precautions During Handling:

Technical Measures: Ventilation of the place, local exhaust of vapours

(in case of product decomposition).

Precautions: Avoid contact with eyes and skin and breathing of any activated

chlorine dioxide vapours.

Wear personal protective equipment, maintain eye washer/shower

facilities and source of running water in the vicinity.

Safe Handling Advice: Handle product with care and avoid contamination.

7.2 Precautions During Storage:

Storage Conditions: Store in a cool, clean, well ventilated area.

Do not store on wooden surfaces or flammable pallets.

Keep away from incompatible (especially acids) and combustible materials,

from direct sunlight and heating sources.

Provide water facilities, do not allow spilt product dry.

7.3 Packaging Materials: Do not use common steel, aluminium, copper and its alloys or rubber.

Use stainless steel, glass, ceramics, polyethylene, PVC.

VIII. Exposure Controls/Personal Protection

8.1 Special Protective Measures

Respiratory: Under normal conditions of use / handling no respiratory protection is

required. Should chlorine dioxide gas be released, ensure working room air concentrations are less than 0.1 ppm chlorine dioxide gas S.C.B.A.

when chlorine dioxide gas is evolved.

Hand: Wear plastic gloves. Eye: Wear goggles.

Skin: Work suit, preferably made of PVC, Neoprene, nitrile rubber.

Avoid leather, cotton or natural rubber due to fire risk.

8.2 Exposure Limits: For chlorine dioxide gas:

0.1ppm (0.3mg/cu.m) 0.3ppm (0.9mg/cu.m) STEL

IX. Physical and Chemical Properties

9.1 Physical State: Liquid.

9.2 Temperature Characteristics: Freezing point: - 10°C

9.3 Solubility:

In Water: Completely miscible.

In Solvents: Not soluble.

9.4 Flammability: Not combustible

X. Stability and Reactivity

10.1 Stability: Minimum 6 months in unopened containers.

10.2 Conditions/Materials to Avoid: Avoid exposure to direct sunlight and heat.

Decomposed by acids and organic matter.

10.3 Decomposition Temperature

and Dangerous Products

Released:

At very high temperatures i.e. greater than 100°C, it may decompose to produce chlorine dioxide gas which can cause overpressure and burst in confined spaces. Toxic chlorine dioxide gas maythen be released.

XI. Toxicological Information

11.1 Acute Toxicity: LD50 orl.rat (by feeding-m/f): >4000 mg/kg.

LD50 dermal.rbt: => 5000 mg/kg.

11.2 Local Effects: Skin contact: Not irritating (patch test, rabbit, 4h)

Eyes: Irritant to eyes (Draiza test rabbit).

11.3 Long Term Toxicity: Carcinogenicity: No carcinogenic potential.

Teratogenicity No teratogenic potential.

XII. Biological Information

12.1 Mobility: Will disperse through aqueous systems.

12.2 Persistence and Degradability: Will degrade.

12.3 Bioaccumulative Potential: Not known.

XIII. Disposal Considerations

13.1 Disposal of product: Disposal of through approved waste disposals operative.

13.2 Disposal of packaging: Rinse empty containers thoroughly before disposal.

XIV. Transport Information

14.1 UN No: 1908

14.2 UK Road:

Hazchem Code: 2X (additional advice A)
Classification: Corrosive substance, Class 8.

Packing Group: II.

14.3 Spillage: Larger than 25 litres, decontaminate with sodium sulphite solution with a

contact time of least 10 minutes, then dilute with water & flush to foul drain.

XV. Regulatory Information

15.1 The chemicals (Hazard Information and Packaging for Supply) Regulations 1994:

Index No : Not Listed

Risk Phrases: R8 Contact with combustible materials may cause fire

R22 Harmful if swallowed

R32 Contact with acids liberates very toxic gas

Safety phrases: S14 Keep away from acids

S17 Keep away from combustible materials

S26 In case of contact with eyes, rinse immediately with plenty of

water and seek medical advice.

S36 Wear suitable protective clothing, gloves and eye/face protection.

/37

Classifications Symbols: O Oxidiser

Xn Harmful

XVI. Other Information

16.1 Recommended uses: Industrial Odour Control Compound

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Always READ material safety data sheet before use.

preserving the environment

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