

#### I. Product and Company Identification

Identification of the substance or preparation:

#### PERSNICKETY DIOXIDE

1.2 Company Identification: Odour Control Systems Limited

33A Castle Close,

Hawarden Industrial Park

Manor Lane, Hawarden, Deeside. CH5 3PP.

Tel: 01244 536700 Fax: 01244 535184

#### П. Composition/Information on Ingredients

Chemical Composition: Preparation of proprietary oxidising ingredients including Aqueous 2.1

solution of stabilised buffered chlorine dioxide, polymers, surfactants,

terpenes and inert ingredients.

2.2 Risk Phrases: R8 Contact with combustible materials may cause fire.

> R22 Harmful if swallowed.

R32 Contact with acids liberates toxic gas.

2.3 Classification/Symbol: 0 Oxidising

Harmful XN

#### Hazards Identification III.

3.1 Adverse Health Effects: Irritating to eyes.

Harmful if swallowed.

3.2 Environmental Effects: Toxic to aquatic organisms.

Physical & Chemical In contact with acids or when heated or under sunlight, may develop very

toxic gas (chlorine dioxide) which may cause risk of explosion.

Contact with combustible materials (grease, fats, wood, cellulose,

paper, etc) may cause fire.

Contact with reducing agents and sulphur containing substances

causes violent exothermic reaction.

#### IV. First Aid Measures

4.1 Eye Contact: Immediately flush eyes with water for at

least 15 minutes.

4.2 Skin contact: Remove contaminated clothing and wash affected areas with soap & water.

Soak contaminated clothing with water to prevent fire risk.

Seek medical attention.

4.3 Inhalation: Remove to fresh air. Seek medical attention. (If the patient is affected by

chlorine dioxide, this is very toxic and requires urgent medical attention).

4.4 Ingestion: Rinse mouth with water and give water or milk to drink.

Seek urgent medical attention.

#### V. Fire Fighting Measures

extinguishing media

5.1 Special Fire/Explosion May decompose in fire producing toxic chlorine compounds.

Strong oxidising agent will assist combustion.

5.2 Products of combustion Not combustible. Toxic chlorine compounds released in fire.

5.3 Fire Fighting procedures/ Wear S.C.B.A. for chlorine/chlorine dioxide

Keep containers cool with water spray. Avoid dispersion in the water courses.

Do not use carbon dioxide or organic material.

Risk of drums bursting.

#### VI. Accidental Release Measures

6.1 Personal Protection: Wear goggles, plastic gloves (not rubber) and boots.

Eyewash facilities should be available.

6.2 Environmental Precautions Do not allow to dry. If possible drench with water.

Contain with inert material.

Pump into a suitable container or otherwise absorb in sand. Do not absorb in sawdust or other combustible materials.

If substance has entered a water course or sewer or contaminated

soil or vegetation, advise police.

6.3 Methods of Cleaning Up: Flush away any residues with excess water.

#### VII. Handling and Storage

7.1 Precautions During Handling:

Technical Measures: Ventilation of the place, local exhaust of vapours

(in case of product decomposition).

Precautions: Avoid contact with eyes and skin and breathing of any activated

chlorine dioxide vapours.

Wear personal protective equipment, maintain eye washer/shower

facilities and source of running water in the vicinity.

Safe Handling Advice: Handle product with care and avoid contamination.

7.2 Precautions During Storage:

Storage Conditions: Store in a cool, clean, well ventilated area.

Do not store on wooden surfaces or flammable pallets.

Keep away from incompatible (especially acids) and combustible materials,

from direct sunlight and heating sources.

Provide water facilities, do not allow spilt product dry.

7.3 Packaging Materials: Do not use common steel, aluminium, copper and its alloys or rubber.

Use stainless steel, glass, ceramics, polyethylene, PVC.

#### VIII. Exposure Controls/Personal Protection

8.1 Special Protective Measures

Respiratory: Under normal conditions of use / handling no respiratory protection is

required. Should chlorine dioxide gas be released, ensure working room air concentrations are less than 0.1 ppm chlorine dioxide gas S.C.B.A.

when chlorine dioxide gas is evolved.

Hand: Wear plastic gloves.

Eye: Wear goggles.

Skin: Work suit, preferably made of PVC, Neoprene, nitrile rubber.

Avoid leather, cotton or natural rubber due to fire risk.

8.2 Exposure Limits: For chlorine dioxide gas:

0.1ppm (0.3mg/cu.m)

0.3ppm (0.9mg/cu.m) STEL

#### IX. Physical and Chemical Properties

9.1 Physical State: Pale yellow-green solution

9.2 Odour: Ozone like

9.3 Temperature Characteristics Boiling point : < 214 deg F

Freezing point :< minus 4 deg C.

9.4 pH: 10.00 – 10.5

9.5 Solubility:

In water Completely miscible

In solvents Not soluble.

9.6 Vapour pressure Similar to water

9.7 Density: 1.06

9.8 Flammability Not combustible

### X. Stability and Reactivity

10.1 Stability: Minimum 6 months in unopened containers.

10.2 Conditions/Materials to Avoid: Avoid exposure to direct sunlight and heat.

Decomposed by acids and organic matter.

10.3 Decomposition Temperature

and Dangerous Products

Released:

May decompose to produce chlorine dioxide gas which can cause

overpressure and burst in confined spaces.

Toxic chlorine dioxide gas may then be released.

### XI. Toxicological Information

11.1 Acute Toxicity: LD50 orl.rat (by feeding-m/f): 2.5 gm kg.

Category III LD50 dermal.rbt : => no dermal irritation on prolonged contact.

11.2 Local effects Skin contact: not irritating (patch test, rabbit, 4h)

Eyes: mild irritant to eyes (Draiza test rabbit).

11.3 Sensitisation: No data.

11.4 Chronic Toxicity: Acute, subacute and chronic (organs & systems) effects.

Acute subacute and chronic toxicity tests in mice by oral feeding were performed with 5% stabilised chlorine dioxide in several dilutions.

Conclusions:

A) Undiluted 5% stabilised chlorine dioxide was toxic to mice when fed

0.5ml directly from gavage.

B) Dilutions ranging from 1:10 to 1:240 were found to be non toxic when fed by stomach trocar.

C) Chronic feeding tests by gavage and in drinking water of 5% stabilised chlorine dioxide solution was non-toxic.

11.5 Long term toxicity: Carcinogenicity; No carcinogenic potential.

11.6 Experience in humans: No significant effect after application in drinking water for up to 12 weeks.

Continued inhalation of decomposition products may cause lung oedema.

#### XII. Biological Information

12.1 Mobility: Will disperse through aqueous systems.

12.2 Persistence and Degradability: Will degrade.

12.3 Bioaccumulative Potential: Not known.

### XIII. Disposal Considerations

13.1 Disposal of product: Disposal of through approved waste disposals operative.

13.2 Disposal of packaging: Rinse empty containers thoroughly before disposal.

### XIV. Transport Information

14.1 UN No: 3266

14.2 UK Road:

Hazchem code:

Classification: 8 (Corrosive)

Packing group:

14.3 Spillage: Larger than 25 litres, decontaminate with sodium sulphite solution with a

contact time of at least 10 mins, then dilute with water & flush to foul drain.

### XV. Regulatory Information

15.1 The chemicals (Hazard Information and Packaging for Supply) Regulations 1994:

Index No : Not Listed

Risk Phrases: R8 Contact with combustible materials may cause fire

R22 Harmful if swallowed

R32 Contact with acids liberates very toxic gas

Safety phrases: S14 Keep away from acids

S17 Keep away from combustible materials

S36/ Wear suitable protective clothing, gloves and eye/face protection.

37/39

Classification Symbols: O Oxidizer

Xn Harmful

15.2 Exposure Limits: For chlorine dioxide gas:-

0.1 ppm (0.3 mg/cu.m).

0.2 ppm (0.9 mg/cu.m). STEL.

#### XVI. Other Information

16.1 Recommended uses: Industrial Odour Control Compound

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# preserving the environment



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